

Spontaneous Reheating during Crystallization of Stardust: Resolution of an Interstellar Medium Paradox

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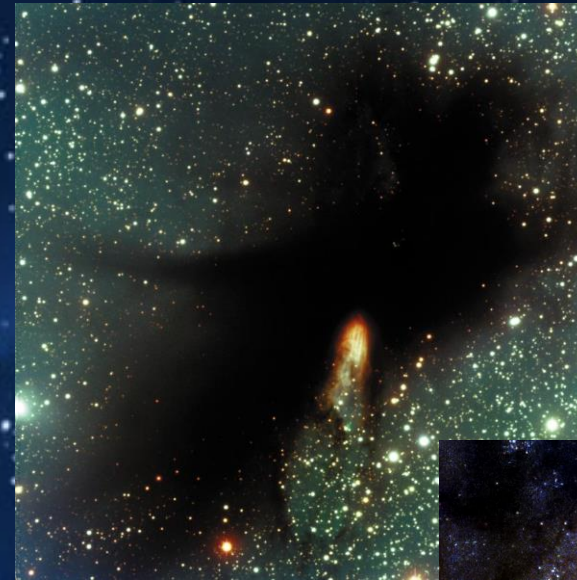
² NASA Ames Research Center

Dust in space is everywhere

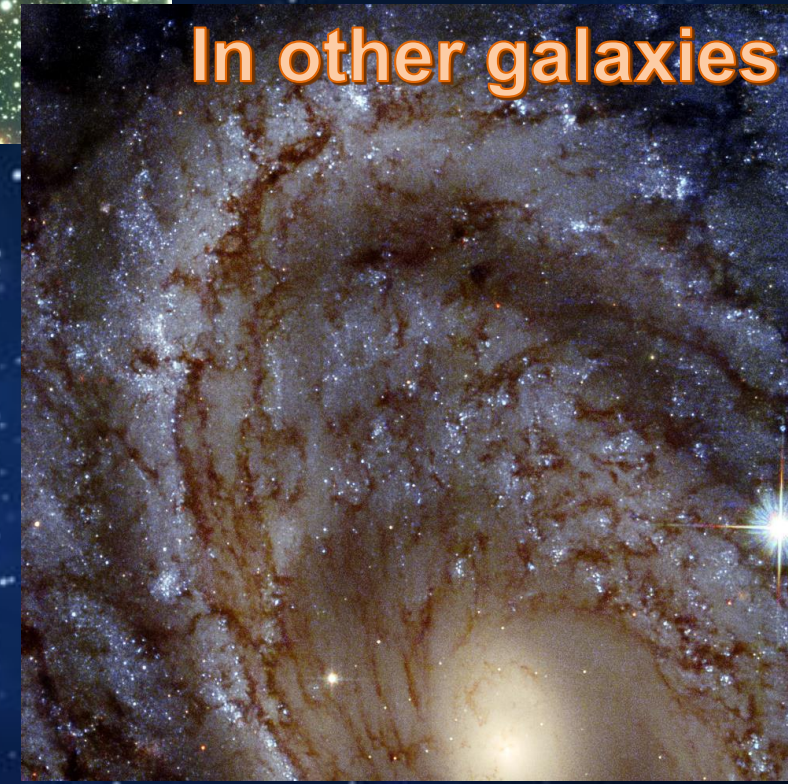
Star birth



In between stars



In other galaxies



Star death



Crystal structure matter



GRAPHITE

Graphite



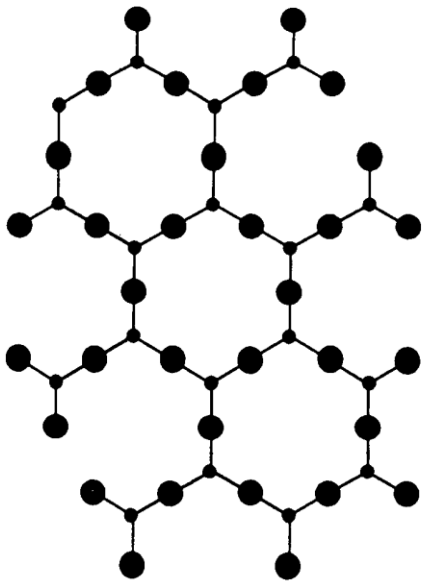
**Amorphous
Carbon**

Diamonds

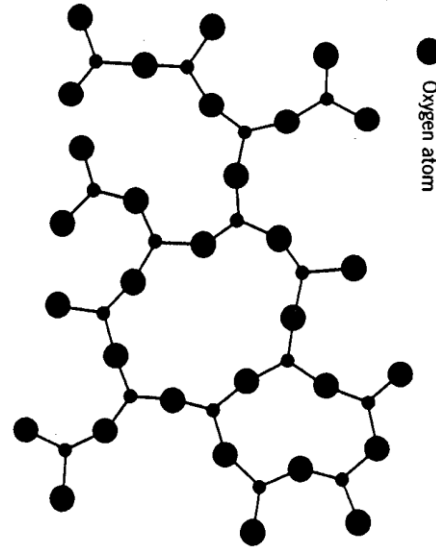


Crystalline vs. amorphous dust

crystalline



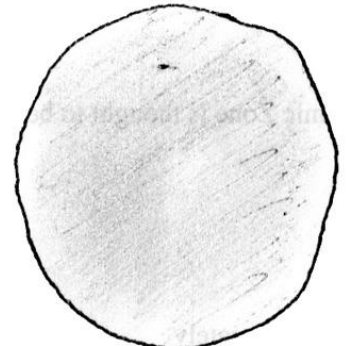
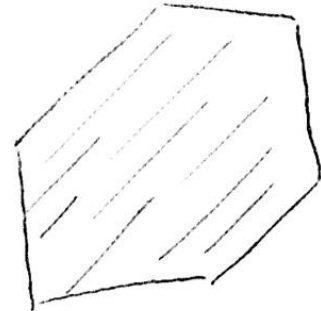
non-crystalline



● Silicon atom
● Oxygen atom

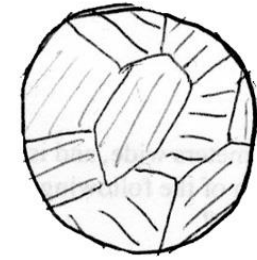


Single Crystal
(ordered, anisotropic)



Glassy Grain
(disordered, isotropic)

Polycrystalline
Grains



monomineralic



polymineralic



polymineralic
including glass



Results:



As molten rock cools it nucleates crystals

The formation of the crystals releases heat

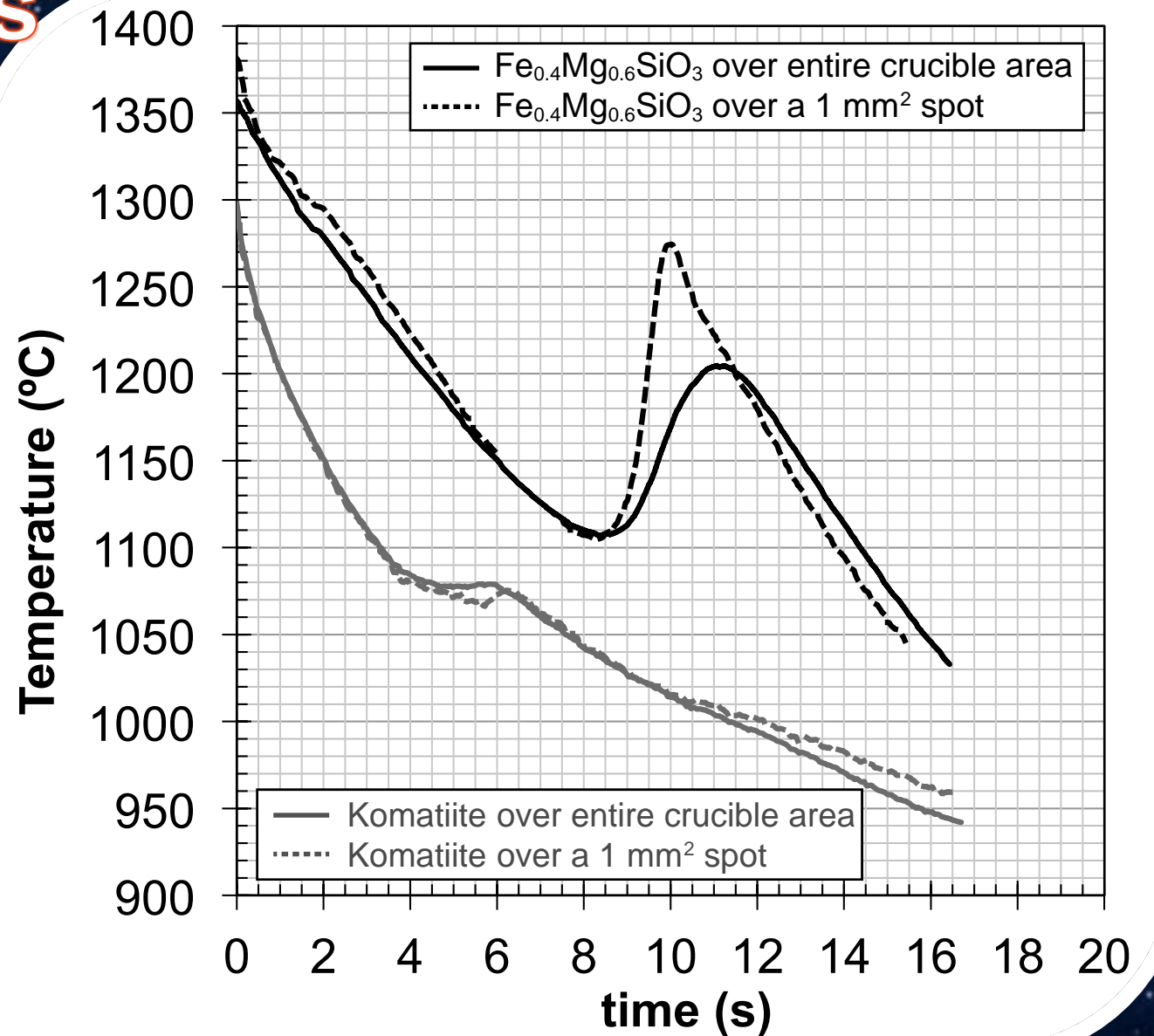
The crystals glow brightly compared to the surround liquid

Experimental Results

We studied the heating and cooling of mineral samples

$\text{Fe}_{0.4}\text{Mg}_{0.6}\text{SiO}_3$ is a similar to silicate space dust

On cooling, the material spontaneously heats up!



Energy release from crystallization

At ~1000 K curve goes -ve

Sample is generating heat from crystallization

→ Shines brighter than surrounding

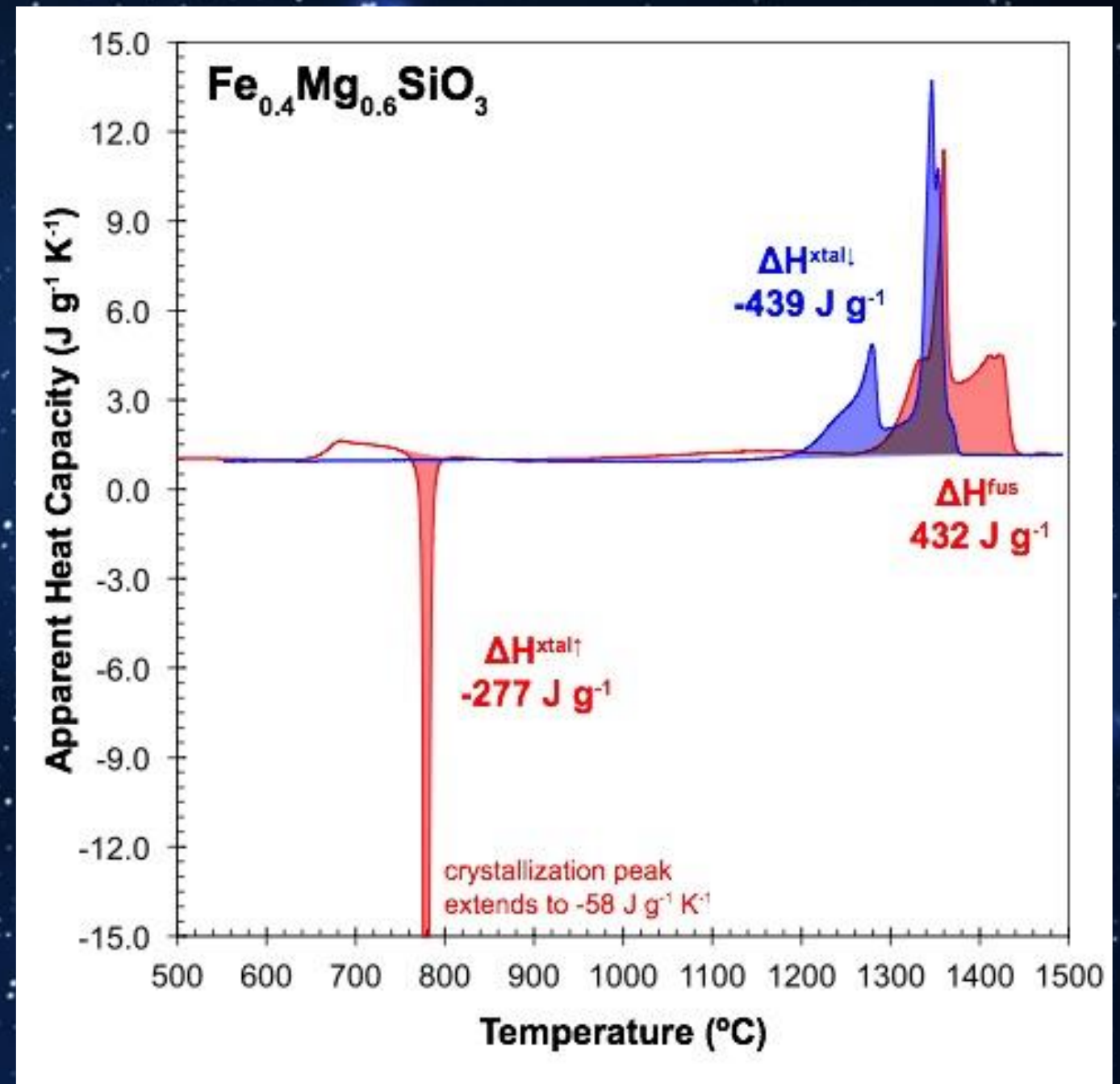
Brightness is proportional to T^4

Small change in T → big change in brightness

For space dust – grains at ~1000K may be able to crystallize

→ heat up

→ Shine more brightly than surrounding amorphous dust



Implications/Summary

- As dust forms it can crystallize
- The dust heats → Brightens → easier to see
- Once it cools, crystalline features will be hidden.
- Dust forming around stars will show crystalline dust,
- Cool dust in the ISM will not.
- Solve the paradox
- Whether dust is crystalline or amorphous affects **EVERYTHING!**

